

THERMAL MOTION IN SODIUM NITRITE: THE TEMPERATURE DEPENDENCE OF DEBYE-WALLER FACTORS

Y. Kawamura*, N. Nitta, H. Kasano and H. Mashiyama

*Department of Physics, Faculty of Science, Yamaguchi University,
Yamaguchi 753-8512, Japan*

(Received 2 September 2001)

In order to understand the relation between the transition mechanism and the thermal vibration, the temperature dependence of the Debye-Waller factor is investigated in NaNO_2 by single-crystal X-ray diffractometry. The crystal parameters are refined by least squares calculations both in paraelectric and ferroelectric phases. The maximum entropy method is also employed to estimate the spontaneous polarization. The Debye-Waller factor displays a minor anomaly at the order-disorder transition point. The result is compared with K_2SeO_4 , a typical displacive crystal.

Keywords: Debye-Waller factor; order-disorder/displacive type; sodium nitrite; structure analysis; least squares calculation; maximum entropy method

1. INTRODUCTION

Two types of transition mechanism have been recognized in dielectric crystals; the order-disorder type and the displacive type.^[1] From a view point of crystal structure analysis, a split-atom method is successfully applied to the former, while an atom has a definite position with the anisotropic thermal parameters in the latter.

The thermal motion is represented by Debye-Waller (D-W) factors in the structural analysis, where the temperature factor is represented by

$$T = \exp\{ -2\pi^2 [U_{11}(ha^*)^2 + U_{22}(kb^*)^2 + U_{33}(lc^*)^2 + 2U_{12}ha^*kb^* + 2U_{13}ha^*lc^* + 2U_{23}kb^*lc^*] \}. \quad (1)$$

* Corresponding author, e-mail: b4342@sty.cc.yamaguchi-u.ac.jp

The square root of U_{jj} represents the effective amplitude of the thermal vibration along the j -axis of the crystal.

It has been sometimes stated that the D-W factor behaves differently in order-disorder and in displacive type crystals.^[2,3] Figure 1 shows a schematic behavior of the D-W factor. In the former, the D-W factor (marked by closed circles U_2) drops steeply at the structural transition. In the latter, the D-W factor (open circles U_3) would have a cusp-type anomaly at the transition.

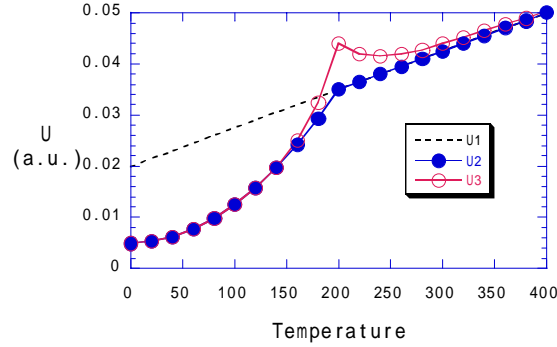


Figure 1. Schematic behavior of the Debye-Waller factor. The Transition is presumed at $T=200$.

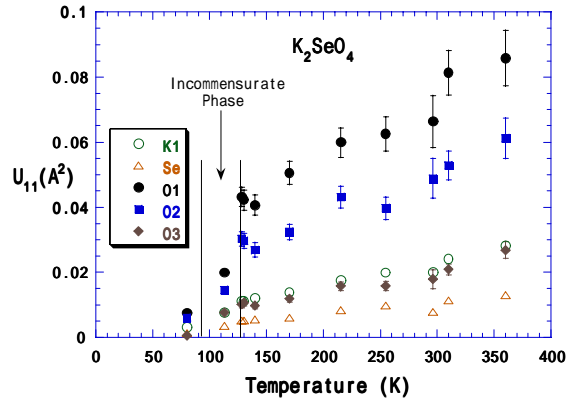


Figure 2. The Debye-Waller factor U_{11} of K_2SeO_4 . The data points at 115K and 80K are after Yamada *et al.*(1984).^[5, 6] The *esd*'s are shown by bars.

Although the conjecture sounds plausible, we have found that the D-W factors of K_2SeO_4 , a typical incommensurate crystal, which has a soft phonon mode, decrease monotonically with decreasing temperature, and behave just as closed circles of Fig. 1. If we choose the space group of K_2SeO_4 as $Pm\bar{c}n$, then the dominant displacement is along the

a -axis. The temperature dependence of the D-W factor U_{11} is shown in Fig. 2.^[4], although the structural transition at 128K is the displacive type.

Since our result is in contradiction to the previous conjecture, we investigate the temperature dependence of D-W factor in NaNO₂, a typical incommensurate crystal that performs the order-disorder type transition.^[7] At room temperature the space group is $Im2m$ ($Z=2$). The spontaneous polarization vanishes at 437K, above which the incommensurate phase exists in a narrow temperature range (less than 2K).^[8] In the high-temperature paraelectric phase, the NO₂ molecule takes the up-down orientated state with equal probability to ascertain the mirror symmetry.

2. EXPERIMENT AND RESULT

Small pieces of NaNO₂ were cut from the transparent part of single crystals. Each specimen was sealed in a glass capillary to avoid moisture and was mounted on a four-circle X-ray diffractometer. Atomic parameters including D-W factors were refined by least squares calculations. A split-atom method was employed to represent the up-down orientated state in the paraelectric phase for all constituent atoms.

Figure 3 shows the temperature dependence of D-W factors for three atoms. The previously reported parameters by several authors are also plotted in the figure; 30K,^[9] 120K,^[10] 295K,^[11] 338K,^[12] 378K,^[12] 418K,^[12] 443K,^[13] 458K^[14] and 498K^[14]. Other 26 temperature points are obtained for 4 samples by our analysis. The temperature region of the incommensurate phase is excluded. Our results are in coincidence with the previous ones essentially. The large amplitude motion of the NO₂ molecule normal to the molecular plane, U_{11} , suggests that the up-down motion of the dipole moment takes place through the rotation of the NO₂ molecule about an axis parallel to the c -axis.^[13]

The scatters at room temperature indicate a sample dependence. In the paraelectric phase, the correlation between the split positions and the thermal parameters induces some scatters in U_{jj} . However, the temperature dependence can be expressed as the following; (1) U_{11} and U_{33} vary with temperature continuously through the transition. (2) U_{22} behaves as U_2 of Fig. 1; a small kink can be recognized at the transition.

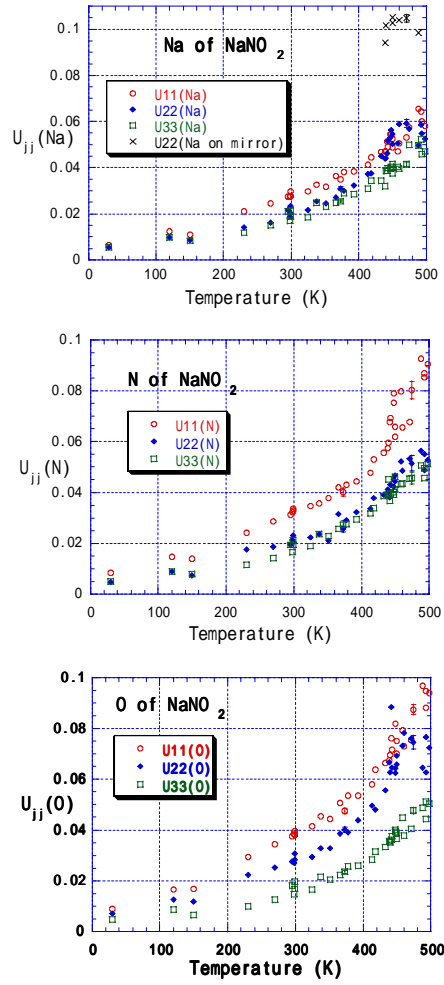


Figure 3. The temperature dependence of the Debye-Waller factors of NaNO_2 .

In the paraelectric phase, the Fourier synthesis shows double peaks for N and elongated peaks for O and Na, which indicates disordered nature. If the atomic position of Na is fixed on the mirror plane, then U_{22} becomes about twice of the split-atom as shown by crosses in the top of Fig. 3.

3. DISCUSSIONS

As mentioned in the introduction, the plausible arguments claimed that U_2 and U_3 of Fig. 1 corresponded to order-disorder and displacive types,

THERMAL MOTION IN SODIUM NITRITE

respectively. Our analysis for NaNO_2 shows that U_{22} , order-disorder in character, is the type U_2 and that U_{11} and U_{33} are almost continuous through the transition. The displacive crystal, K_2SeO_4 , also showed the behavior of the type U_2 .

Some years ago, a theoretical calculation was performed to demonstrate the temperature dependence of the D-W factor; the types of U_3 in Fig. 1 is realized in order-disorder type transitions, while U_2 in displacive type transition.^[15] Our results for displacive type crystal coincide with the theoretical prediction. If the local potential of an atom would not change at the order-disorder transition, the D-W factor would decrease almost continuously around the transition. If the split-atom method was not applied, the D-W factor was the type of U_3 .

In addition to this conclusion, we discuss two points; the order parameter and the spontaneous polarization. Just below the transition, the constituent atoms are not fully ordered; there remains the fraction of split atoms in the ferroelectric phase. So we determined the occupation probability of split atoms as a fitting parameter in the least squares calculations. The result coincided with those obtained by Gohda *et al.*^[12] If we ignored the fraction of split atoms, then U_{22} of N became more than twice larger than that of Fig. 3 just below the transition temperature.

Finally we calculate the spontaneous polarization in ferroelectric phase. By means of least squares calculations, we determine the center of charge density of each atom. If we assigned the net charges for Na, N and O as +1, +3 and -2, then the calculated polarization was $78\mu\text{C}/\text{cm}^2$. But this is much larger than $12\mu\text{C}/\text{cm}^2$ estimated from the

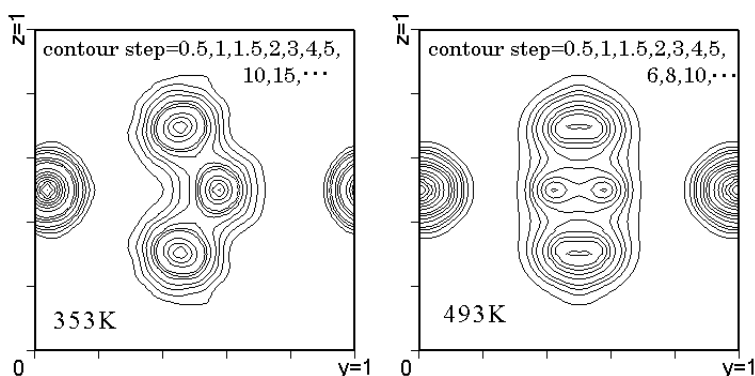


Figure 4. Electron density ($x=1/2$) of NaNO_2 in ferroelectric phase (353K) and in paraelectric phase (493K).

direct measurement of the polarization.^[16] Gohda *et al.* have recently discussed this problem by using multi-pole type electron density function and they have got $7.8\mu\text{C}/\text{cm}^2$.^[9] This method assumed additional fitting parameters in the least squares calculations. We adopt another method; the maximum entropy method is superior, because no *a priori* assumption has been made on charge distribution.^[17] The electron charge is determined by maximum entropy method at 353K, as given in Fig. 4. The net charge is estimated 1.08, 0.16 and -0.62 for Na, N and O, respectively. If the positions of nuclei are those of obtained by the usual least squares calculations, then the polarization is $18.1\mu\text{C}/\text{cm}^2$, much closer to experimental value of the polarization.

Since the bond length N-O is 1.251(2)Å at 353K, the empirical method of the bond valence sum^[18] predicts the bond valence to be 1.35(4). If this charge is transferred from O^{2-} to N^{3+} , then the net charge of O becomes -0.65(4) in agreement with our estimation.

References

- [1] M. E. Lines and A. M. Glass, *Principles and Applications of Ferroelectrics and Related Materials* (Clarendon, Oxford, 1977).
- [2] K. Itoh, H. Matsunaga and E. Nakamura, *Phys. Status Solidi* **97**, 289 (1980).
- [3] K. Itoh, *J. Phys. Soc. Jpn.* **53**, 2049 (1984).
- [4] N. Nitta, *Master's Thesis* (Yamaguchi Univ., 2001, unpublished).
- [5] N. Yamada and T. Ikeda, *J. Phys. Soc. Jpn.* **53**, 2555 (1984).
- [6] N. Yamada, Y. Ono and T. Ikeda, *J. Phys. Soc. Jpn.* **53**, 2565 (1984).
- [7] S. Sawada, S. Nomura, S. Fujii and I. Yoshida, *Phys. Rev. Letters* **1**, 320, (1958).
- [8] S. Tanisaki, *J. Phys. Soc. Jpn.* **16**, 579 (1961).
- [9] T. Gohda, M. Ichikawa, T. Gustafsson and I. Olovsson, *Acta Crystallogr. B* **56**, 11 (2000).
- [10] M. Okuda, S. Ohba, Y. Saito, T. Ito and I. Shibuya, *Acta Crystallogr. B* **46**, 343 (1990).
- [11] T. Gohda, M. Ichikawa, T. Gustafsson and I. Olovsson, *J. Korean Phys. Soc.* **32**, S189 (1998).
- [12] T. Gohda, M. Ichikawa, T. Gustafsson and I. Olovsson, *Phys. Rev.* **B63**, 14101 (2000).
- [13] K. Komatsu, K. Itoh and E. Nakamura, *J. Phys. Soc. Jpn.* **57**, 2836 (1988).
- [14] M. I. Kay, J. A. Gonzalo and R. Maglic, *Ferroelectrics* **9**, 179 (1975).
- [15] K. Fujii and T. Matsubara, *Prog. Theor. Phys.* **78**, 177 (1987).
- [16] K. Hamano, *J. Phys. Soc. Jpn.* **35**, 157 (1973).
- [17] S. Kumazawa, Y. Kubota, M. Takata and M. Sakata, *J. Appl. Cryst.* **26**, 453 (1993)
- [18] I. D. Brown and D. Altermatt, *Acta Cryst.* **B41**, 241 (1985).